# Physical Vapor Deposition of Bronze-Route Nb<sub>3</sub>Sn for SRF Cavities

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#### Why investigate bronze routes?

- Move reaction window from ~1100°C to ~700°C
  - Avoid the Nb<sub>6</sub>Sn<sub>5</sub> and NbSn<sub>2</sub> phases by exploiting ternary Cu-Sn-Nb system
- Make the reaction compatible with cavity bodies made from copper; avoid bulk niobium (and)
  - Possible cost and formability advantages
  - Potential direct application with conduction cooling
    - Genesis of ideas traces back to founding of IARC at Fermilab
- Avoid Sn vapor, chlorides of Sn and Nb polishing chemistry (corrosive, toxicity).
- Tap into wealth of processing knowledge from composite wires.
- If desired, can push grain size to < 50 nm; very low roughness.



# Cu-Sn-Nb pathways for SRF

- Apply Nb to bronze and convert entirely to Nb<sub>3</sub>Sn
  - Also works for bronzed copper  $\rightarrow$  utilize Cu cavities
  - More complicated geometries are possible, e.g. with diffusion barriers
- Apply bronze to Nb and react to form Nb<sub>3</sub>Sn, then remove bronze post-process
  - Utilize Nb/Cu bimetal, e.g. hydroformed
    Nb/Cu cavities
  - Use Nb cavities





"Bronze" need not be bulk bronze (although bulk bronze is an easy starting point for studies) and can be bronzed Cu

#### First studies: Nb on $\alpha$ -bronze

- Substrates: Mechanically polished zero phosphorous Cu-15%wt Sn and Cu-15%wt Sn- 0.3%wt Ti bronzes.
  - Commercial Cu and bronze usually has P as a deoxidizing agent. Ti was explored here as a getter.
- Substrates were cold-rolled and homogenized at 525°C prior to polishing.
- Nb film deposition conditions.
  - Background pressure ~5E-9 Torr
  - Processing gas 8 mTorr Ar
  - Deposition rate 22.4 nm per minute
  - Niobium film  $T_c$  9.2 K,  $\Delta T_c$  0.1 K
- Both process routes took place in the deposition chamber; no vacuum break or exposure to ambient





#### Nb deposited on hot bronze achieves 10x faster reaction



14-16 K  $T_c$  transition consistent with CTE mismatch between Nb<sub>3</sub>Sn and bronze. Also seen by CERN for direct deposition of Nb<sub>3</sub>Sn on copper



#### Nb on bronze + post-reaction behaves like processes in wires



Similar effect of CTE mismatch between Nb<sub>3</sub>Sn and bronze as for route 1



# Columnar grain structure produced by route 1 was never observed in bronze route

- Columnar grains through entire thickness
- Pipeline diffusion at bronze
  GBs → larger Nb<sub>3</sub>Sn grains
- Bronze twin boundaries are evident beneath Nb<sub>3</sub>Sn
- Very low roughness  $R_{\rm A} \sim 7-10$  nm in 100  $\mu$ m<sup>2</sup>.
- Sn content of Nb<sub>3</sub>Sn layer is measured (EDX) at 25% or higher in both substrates.





#### Post-deposition reaction produces structures like those seen in wires

- Tendency toward high-angle equiaxed grains
- Evidence for pipeline diffusion along Ti alloyed bronze GBs, with enriched Sn and Ti.
- Very low roughness  $R_{\Delta} \sim 10 15$  nm in 100  $\mu$ m<sup>2</sup>.
- Sn content of Nb<sub>3</sub>Sn layer
- 22–23.5 % no Ti
- 24.5–25.5 % with Ti
- Large Ti<sub>6</sub>Sn<sub>5</sub> grains in films on Ti alloyed bronze.

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#### **Discussion and summary**

- Address the CTE mismatch issues, e.g. by engineering the copper
  - But 15 K is already sufficient to use conduction cooling for low  $R_{BCS}$
- The hot bronze method may be better
  - Full tin content in  $Nb_3Sn$
- What is the significance of the unique grain structure obtained by deposition of Nb onto hot bronze?
  - − Hot bronze in vacuum has a tin-rich surface
    → high tin activity. Can this be exploited further?
  - Can Nb deposition properties give microstructure control?
    - $\rightarrow$  HIPIMS and biasing schemes are available and not yet fully exploited



#### **Discussion and summary**

- While roughness is extremely low inside the underlying bronze grains, twin boundaries and grain boundaries affect the Nb<sub>3</sub>Sn coating. Is it important to suppress twins and GB effects?
  - Alloying with Ti and other elements increases stacking fault energy, reduces twinning
- What is the significance of the very low roughness? Does this trade off with very high GB density?
- RF measurements are needed.
- Copper in GBs studies are needed.



#### Acknowledgment

- Many thanks to Akihiro Kikuchi (NIMS, Japan) for providing Ti alloyed bronze.
- Alexander Wozny and Jonathan Wozny for their support with bronze substrate polishing.
- This work at ASC, NHMFL-FSU was funded by U.S. Department of Energy, Office of Science, Office of High Energy Physics under Award No. DE-SC 0018379.
- A portion of this work was performed at the National High Magnetic Field Laboratory, which is supported by National Science Foundation Cooperative Agreement No. DMR-1644779 and the State of Florida.



### Backup Slides



# Bronze route Nb<sub>3</sub>Sn films for SRF cavities

Nb<sub>3</sub>Sn

Nb

bronze

at 200 °C

Cu-Sn layer

etch

- Low temperature deposition allows bronze or Cu base cavity.
- Simplest case, Nb<sub>3</sub>Sn coated bronze cavities.
- Significant material cost saving from switching Nb host cavity to a bronze or Cu host cavity.
- Easy fabrication and scaling.
- Cu at grain boundaries, Effect on RF?
- Low thermal conductivity, Engineering solutions?





#### Proposed deposition chamber



# More introduction

Nb – Sn phase diagram

**Bronze process** 

LIQUID 2000 2000 NbaSn + LIQUID Nb a S 15004 ≪-NIOBIUM AND Nb35n 1000 930 ± 8°C boSns+ LIQUID 845 ± 7°C Nb6Sn5 500 500 NbSng + LIQUID Nb<sub>6</sub>Sn<sub>5</sub> + Nb Sn<sub>2</sub> 231.9°C Nb3Sn+Nb6Sn5 N b 5 n<sub>2</sub> + 5 n J. P. Charlesworth et al J. Mat. Sci.5, 580 (1970) Nb<sub>3</sub>Sn layer at the Nb core interface lower Diffusion reaction  $\alpha$  bronze heat treatment

#### Fractographs of a bronze route Nb<sub>3</sub>Sn filament



#### To synthesis pure Nb<sub>3</sub>Sn by reacting Nb and Sn directly require temperatures above 930 °C.

- Eliminates the use of Cu substrate cavity in Nb – Sn direct reaction -> Significant material costs
- Require sophisticated furnace systems for Nb – Sn reaction.
- Bronze route is a well-established technique in Nb<sub>3</sub>Sn wire fabrication.
- Bronze route guarantees only the pure  $Nb_3Sn$  phase at much lower temperatures (~600 800 °C).



P.J. Lee, D.C. Larbalestier, IEEE Transactions 11 (2001)

15

#### Cu-Sn phase digram





#### CTE mismatch



